

WORKSHEET 3
CHEMISTRY

Name _____
last first

Set-ups must be shown where applicable. You will not receive credit for only answers shown
Problem sets are due within the first five minutes of lecture on the due date.
No late work will be accepted.

1. Give the correct answers for the following:

- | | | |
|--|--------|---------------------------------|
| a. Give the symbol for a metalloid in period 4 | Answer | <u>Ge or As</u> |
| b. What is the common (family) name of group IIB | Answer | <u>Transition element/metal</u> |
| c. What group number are the inert gases | Answer | <u>VIII A</u> |
| d. Give the symbol for the alkali metal in period 4 | Answer | <u>K</u> |
| e. Write the symbol for element with the atomic number of 12 | Answer | <u>Mg</u> |
| f. Give the atomic mass of sodium | Answer | <u>23.0 or 22.99</u> |

2. a. Place a {s= solid g = gas l = liquid } next to the elements indicating its physical state at room temperature.

Plus :

b. Place a D next to the diatomic elements

Hg <u>l</u>	Kr <u>g</u>	Cl <u>g,D</u>
Ca <u>S</u>	As <u>s</u>	Si <u>s</u>
Br <u>l,D</u>	I <u>s,D</u>	Ni <u>s</u>

3. Put an "X" in the correct column to classify the following:

	<u>Metal</u>	<u>Nonmetal</u>	<u>Metalloid</u>
a. Magnesium	<u> X </u>	-----	-----
b. Potassium	<u> X </u>	-----	-----
c. Selenium	-----	<u> X </u>	-----
d. Arsenic	-----	-----	<u> X </u>
e. Sulfur	-----	<u> X </u>	-----
f. Lithium	<u> X </u>	-----	-----
g. Manganese	<u> X </u>	-----	-----
h. Boron	-----	-----	<u> X </u>

4. Put an "X" in the correct column to classify the following:

	Element	Compound	Mixture
a. Cherry Kool-Aid	-----	-----	___X___
b. Herb Tea	-----	-----	___X___
c. Br ₂	___X___	-----	-----
d. NH ₃	-----	___X___	-----
e. Chocolate	-----	-----	___X___
f. Air	-----	-----	___X___
g. N ₂	___X___	-----	-----
h. chocolate milk	-----	-----	___X___
i. Si	___X___	-----	-----

5. Indicate whether the following are heterogeneous or homogeneous mixtures:

a. Grape Jelly	<u>HOMOGENEOUS</u>
b. Chocolate chip cookies	<u>HETEROGENEOUS</u>
c. ^{Red} Nail Polish (Red = only color!)	<u>HOMOGENEOUS</u>
d. Oak lumber	<u>HETEROGENEOUS</u>
e. Chicken pot pie	<u>HETEROGENEOUS</u>
f. Paint Thinner (looks like gasoline)	<u>HOMOGENEOUS</u>
g. Burrito	<u>HETEROGENEOUS</u>

6. Indicate whether the following are potential or kinetic energy

a. A can of paint sitting on the edge of a ladder	<u>POTENTIAL</u>
b. A Tidal Wave	<u>KINETIC</u>
c. A 70 lb dog jumping on your lap	<u>KINETIC</u>
d. Stampeding turtles	<u>KINETIC</u>
e. A loaded bow and arrow	<u>POTENTIAL</u>

7. Classify each of the following as either a physical change or a chemical change

Rem.: Chemical changes are frequently accompanied by one or more physical changes

a. Sawing a board into two	<u>PHYSICAL</u>
b. A bear sitting on your lunch	<u>PHYSICAL</u>
c. Snowflakes form	<u>PHYSICAL</u>
e. A nail rusting	<u>CHEMICAL</u>